



## Department of Civil and Environmental Engineering Seminar

### Absorptive and Electrochemical CO<sub>2</sub> Separation Processes

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**Friday January 31, 2020, 12:45-1:45PM @ Bingham 304**

**Lunch and discussion starting at 12:00PM at Bingham 102 (Vose Room)**

#### Abstract

Among the grand challenges that engineers face today is the global climate change. Lessons learned from the industrialization era; 21<sup>st</sup> century engineers seek alternative approaches and processes that are environmentally benign. To mitigate global climate change and reduce carbon emissions from major contributors, discovery of novel materials and processes are essential for the foreseeable future. This talk will present the design approach and utility of ionic liquids (ILs) for capturing and utilizing CO<sub>2</sub> in various applications such as post-combustion. In addition, adaptation of IL-based materials for space applications to remove metabolically generated CO<sub>2</sub> from air will be discussed.

ILs as salts that are in liquid state below 100 °C and some below room temperature. They have wide liquidus range and present negligible vapor pressures with high thermal stability. Therefore, ILs are recognized as 'green solvents'. ILs' high CO<sub>2</sub> solubility and selectivity have attracted interest for separations from post and pre-combustion gases. Their affinity to CO<sub>2</sub> can be tuned by the choice of anion-cation pair and appending functional groups; thus enabling energy savings in separations. Dual utility of ILs as absorbing solvents and electrolytes, owing to their ionic conductivity, further enables the electrochemical separations and conversion of CO<sub>2</sub> to other commodities. We have designed ILs specifically for absorptive and electrochemical processes that are energy efficient compared to amine based solvents – the most mature technology in CO<sub>2</sub> capture. Currently, we are adapting these materials and design principles for the air revitalization needs to remove metabolically generated CO<sub>2</sub> in spacecraft and at the international space station. This presentation will provide performance comparison of our novel encapsulated ILs to the state-of-the-art zeolite absorbers for processes that need to operate under microgravity.



#### About the Speaker

Dr. Gurkan received a B.S. degree in Chemical Engineering from *Middle East Technical University*, (2004), M.S. degree from *University of Toledo* (2006) and PhD from *University of Notre Dame du Lac* (2011). She was a postdoctoral researcher at the *Massachusetts Institute of Technology* and the *University of Akron* before joining *Case Western Reserve University* (CWRU) as an assistant professor in 2016. Her current research interests include CO<sub>2</sub> capture & conversion, fundamentals of electrochemical processes in ionic liquids and deep eutectic solvents in particular for energy storage. Dr. Gurkan received the Best Master Thesis Award from the University of Toledo and was awarded Bayer Pre-doctoral Fellowship at the University of Notre Dame. She recently received the NASA Early Career Faculty and the American Chemical Society's Petroleum Research Fund - Doctoral New Investigator awards. She is the Thrust Leader and the Scientific Integration Officer for the DOE Energy Frontier Research Center: Breakthrough Electrolytes for Energy Storage (BEES) led by CWRU. She enjoys hiking, painting and spending time with her two daughters.